New Alkaloids and a Tetraflavonoid from Cephalotaxus wilsoniana

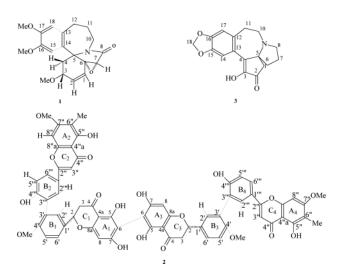
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A new homoerythrina alkaloid, C-3-epi-wilsonione (1), a new tetraflavonoid, taiwanhomoflavone C (2), and a new stereoisomer of desmethylcephalotaxinone (3) have been isolated from the leaves and heartwood of Cephalotaxus wilsoniana, respectively. The structures were elucidated by spectroscopic methods. Compound **1** showed cytotoxic activity against a number of human cancer cell lines in vitro.

Various constituents of Cephalotaxus wilsoniana Hayata (Cephalotaxaceae) have been reported.¹⁻³ During a search for cytotoxic compounds from Formosan plants, we found that a crude CHCl₃ extract of the leaves and heartwood from C. wilsoniana showed potent cytotoxic activity against Hep G2, 212, Hep 3B, HT-29, and HCT 116 cells in vitro. Therefore, we investigated the constituents of the leaves and heartwood of C. wilsoniana and isolated a new homoerythrina alkaloid, C-3-epi-wilsonione (1), a new tetraflavone, taiwanhomoflavone C (2), a new stereoisomer of desmethylcephalotaxinone (3), and four known constituents, C-3-epi-wilsonine, taiwanhomoflavone A, ginkagetin, and apigenin. In the present paper, the structure elucidation and cytotoxicities of the three new compounds (1-3)are reported.



The molecular formula of 1 was determined to be C₂₀H₂₃-NO₅ by HREIMS (m/z 357.1510 [M⁺]), which was consistent with ¹H and ¹³C NMR data (Experimental Section). The UV absorption maxima were similar to those of C-3-epiwilsonine.¹ The IR absorption of **1** implied the presence of a carbonyl (1689 cm⁻¹) moiety. The proton signals of H-1, H-2, H-3, H₂-4, H-15, H-18, MeO-3, MeO-16, and MeO-17 of 1 were similar to those of C-3-epi-wilsonine. However,

the ¹H NMR spectrum of 1 showed six aliphatic and an oxymethine proton signal. The IR absorption band at 1689 ${
m cm^{-1}}$ together with the carbon signal at δ 167.9 confirmed that 1 contains a =N-CO moiety.⁴ The HMBC correlations of H-7/C-8 and H-2/C-6 and the NOESY correlations of H-1/ H-7, H-1/H-2, and H-3/H-4 further confirmed that C-3-epiwilsonione was 1,2-didehydro-6,7-epoxy-3a,16,17-trimethoxverythrinan-8-one (1). The ¹³C NMR assignments of 1 were deduced from ¹H-decoupled, DEPT, and 2D NMR including ¹H⁻¹H COSY, HMQC, HMBC, and NOESY.

Compound **2** was obtained as an amorphous yellow solid. IR absorptions were indicative of hydroxy (3158 cm⁻¹), conjugated carbonyl (1642 cm⁻¹), and aromatic ring (1607 cm^{-1}) groups. It possesses the molecular formula $C_{66}H_{50}O_{20}$, determined from electrospray ionization mass spectrometry (ESIMS) (M + Na + 2H at m/z 1187). The UV spectrum exhibited absorption maxima at 208, 278, and 329 nm. Its ¹H NMR spectrum displayed signals for 18 aromatic protons, two oxymethine and two methylene, characteristic of two flavonone units,⁵ four methoxy, and two methyl groups (Table 1). The ¹H NMR spectrum of 2 in C₅D₅N indicated crowded signals between δ 7.17 and 7.23 and between 7.77 and 7.83, while these signals in CD_3OD appeared between δ 6.76 and 6.82, between 7.30 and 7.60, and between 7.50 and 7.80, respectively, and were separable. Observing the proton signals measured in C₅D₅N and CD₃OD, repectively, and from detailed analysis of a ¹H-¹H COSY NMR experiment, the aromatic protons were identified as four 1,2,3,4,5-pentasubstituted and four 1,3,4trisubstituted aromatic rings. In addition to the above evidence, the HMBC correlations of two H-2"'/C-4"', two H-3"/C-2", C-4", and C-4"a, two OMe-7"/C-7", two Me-6"/ C-6", two H-8"/C-6", C-7", C-8"a, and C-4"a, two H-6""/C-4"", and two H-5""/C-1"", C-3"", and C-4"", NOESY correlations of two Me-6"/OMe-7" and H-8"/OMe-7", and two hydroxy proton signals at δ 13.93 confirmed that **2** contains two 5,4'-dihydroxy-6-methyl-7-methoxy-3'''-substituted flavone moieties. The HMBC correlations of two MeO-4'/C-4', two H-2'/C-4' and C-6', two H-5'/C-1' and C-3', two H-6'/ C-2 and C-2', two H₂-3/C-4, and two H-8/C-4a, C-7, and C-6, and two hydroxy proton signals at δ 12.69 confirmed that 2 contains two 5,7-dihydroxy-4'-methoxy-6,3'-disubstituted flavonone moieties. In addition, the UV spectrum of 2 showed bathochromic shifts upon addition of AlCl₃, NaOAc, and NaOMe. The above results and deshielding of the signals of substituted carbons C-6 of A₁ and A₃, C-3' of B₁ and B₃, and C-3^{*'''*} of B₂ and B₄ rings, caused by the flavanyl substitution, compared to the monomeric flavonone or

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Table 1.	¹³ C NMR and	¹ H NMR Data	for 2 (C_5D_5N)
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С	ring	δ_{C}	$\delta_{ m H}$	С	ring	$\delta_{\rm C}$	$\delta_{ m H}$
flavonone				flavone			
2	C_1	79.6 ^a	5.63 (dd,	2″	C_2	164.4	
	- 1		13.6, 2.8)		- 20		
3	C_1	43.2	3.02 (dd,	3″	C_2	103.4	6.97 (s)
	10.2	17.2, 2.8)	0	02	100.1	0.01 (5)	
			3.43 (dd,	4″	C_2	183.1	
			4	C_2	105.1		
	C	100.0	17.2, 13.6)	A//		100.0	
4	C_1	196.9		4″a	A_2	103.3	
4a	C_1	105.0		5″	A_2	160.8	
5	A_1	163.4		6″	A_2	105.0	
6	A_1	105.6		7″	A_2	166.0	
7	A_1	162.4		8″	A_2	91.4	6.24 (s)
8	A_1	99.7	6.92 (s)	8″a	A_2	162.1	
8a	A_1	155.5		Me-6"	A_2	7.25	2.17
OH-5	A ₁		12.69	OMe-7"	$\tilde{A_2}$	55.9	3.75
1′	B ₁	131.3	12:00	OH-5"	A_2	0010	13.93
2′	\mathbf{B}_{1}	132.5^{b}	8.05 (d,	1‴	\mathbf{B}_2	123.0	10.00
~	\mathbf{D}_1	152.5	2.4)	1	D2	125.0	
o/	р	199.0	2.4)	2‴	р	100.0	7 09 (1
3′ B ₁	D1	123.0		2	B_2	128.8	7.82 (d,
	_				_		2.4)
4′	B_1	159.0		3‴	B_2	122.4	
5′	B_1	111.5	7.19 (d,	4‴	B_2	162.8	
		8.8)					
6′	B_1	128.0	7.68 (dd,	5‴	B_2	116.8	7.22 (d,
-			8.8, 2.4)				8.8)
OMe-4' B ₁	B_1	55.8	3.76	6‴	B_2	128.8	7.81 (dd
	21	0010	0110	0	22	10010	8.8, 2.4)
2	C_3	79.5 ^a	5.63 (dd,	2″	C_4	164.4	0.0, 2.1)
~	C_3	10.0	13.6, 2.8)	~	\mathbf{C}_4	104.4	
n	C	40.0		3″	C	102.4	0.07 (a)
3	C_3	43.8	2.95 (dd,	3	C_4	103.4	6.97 (s)
			17.2, 2.8)		<i>a</i>	100.1	
			3.38 (dd,	4‴	C_4	183.1	
			17.2, 13.6)				
4	C_3	197.0		4a″	A_4	103.3	
4a	C_3	105.0		5″	A_4	163.8	
5	A_3	163.4		6″	A_4	105.0	
6	A_3	105.6		7″	A_4	166.0	
7	A ₃	162.4		8″	A_4	91.4	6.21 (s)
8	A ₃	99.7	6.92 (s)	8″a	A ₄	162.1	0.21 (5)
8a	A ₃ A ₃	155.5	0.02 (3)	оа Me-6″	A_4 A_4	7.25	2.16
		100.0	19.00				
OH-5	A ₃	101.1	12.69	OMe-7″	A_4	55.9	3.61
1′	B_3	131.1	0.00 (1	1′′′′	B ₄	122.8	a aa (1
2′	B_3	132.0^{b}	8.03 (d,	2‴	B_4	128.8	7.77 (d,
			2.4)				2.4)
3′	B_3	123.0		3‴	B_4	122.4	
4′	B_3	158.9		4‴	B_4	162.8	
5′	\mathbf{B}_{3}	111.5	7.19 (d,	5‴	\mathbf{B}_{4}	116.8	7.19 (d,
- 23	0		8.8)	-	7		8.8)
6' B ₃	B_3	128.1	7.66 (dd,	6‴	B_4	128.8	7.78 (dd
U D3	D 3	160.1	8.8, 2.4)	0	D 4	120.0	8.8, 2.4)
OMe-4′	D	EF O					0.0, 2.4)
Owle-4	B_3	55.8	3.76				

^{*a,b*} Values may be reversed.

flavone, and the NOESY correlations of H-2' (B1 ring)/H-2" (B2 ring) and H-2' (B3 ring)/H-2" (B4 ring) suggested that **2** contained three C–C linkages between C-6 (A₁ ring) and C-6 (A3 ring), C-3' (B1 ring) and C-3" (B2 ring), and C-3' (B₃ ring) and C-3''' (B₄ ring) (Table 1).^{6,7} The ${}^{13}\bar{C}$ NMR signals of 2 (Table 1) were assigned by performing ¹Hdecoupled, DEPT, and ¹H, ¹³C COSY correlation experiments and by comparison with those of corresponding data reported in the literature.^{2-4,7,8} The ¹³C NMR spectrum and the presence of significant MS peaks at $m/z 1164 [M + 2]^+$, 595 $[1187 + 2H - 2b]^+$, and 583 $[1164 - a]^+$ also supported the structure **2**. Thus, compound **2** was characterized as di(5,7,4"'-trihydroxy-2,3-dihydro-4',7"-dimethoxy-6"-methyl-3',3"'-biflavanyl)-6,6-tetraflavone. The molecular formula of 3 was determined to be $C_{17}H_{17}NO_4$ by HREIMS (m/z 299.1155 [M]⁺), which was consistent with the ¹H and ¹³C NMR data. The UV, ¹H NMR, and $[\alpha]_D^{23}$ of **3** were similar to those of desmethylcephalotaxinone.⁹⁻¹¹ In the ¹³C NMR spectrum of 3, the chemical shift values were almost identical to those of corresponding data of synthetic desmethylcephalotaxinone except for C-1 to C-7 and C-10.¹⁰ The ^{13}C NMR signals of **3** were assigned by performing ^{1}H -decoupled, DEPT, and ^{1}H , ^{13}C COSY correlation experiments and by comparison with those of synthetic desmethylcephalotaxinone reported in the literature.^{10,11} The NOESY correlation of **3** showed a cross-peak between H_{\alpha}-1 and H_{\alpha}-6. On the basis of the above evidence, **3** was a stereoisomer of desmethylcephalotaxinone.

A microassay for cytotoxicity of the constituents isolated from *C. wilsoniana* was performed using MTT.^{12–14} The cytotoxicities of **1**, **2**, **3**, and taiwanhomoflavone A were tested against a number of human cancer cell lines. Compound **1** showed cytotoxic activity against the Hep G2, MCF-7, Hep 3B, and HT-29 in a concentration-dependent manner with IC₅₀ values of about 52.0, 42.0, 52.0, and 24.4 μ g/mL, respectively.

Experimental Section

General Experimental Procedures. Melting points are reported uncorrected. Optical rotations were obtained on a

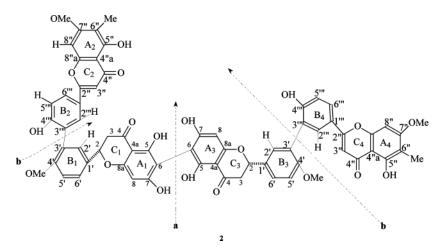


Figure 1. Major MS fragmentation pattern of 2.

JASCO model DIP-370 digital polarimeter; UV spectra were obtained on a JASCO model 7800 UV-vis spectrophotometer; IR spectra were recorded on a Hitachi model 260-30 spectrophotometer; ¹H (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded on a Varian Unity-400 spectrometer, and MS were obtained on a JMS-HX 100 mass spectrometer.

Plant Material. Whole plants of C. wilsoniana were collected at Taichung Hsien, Taiwan, during October 2000. A voucher specimen has been deposited in the author's laboratory

Extraction and Isolation. The air-dried leaves (1.04 kg) of C. wilsoniana were extracted with CHCl₃ at room temperature. The CHCl₃ extract (35 g) was chromatographed over silica gel, and elution with CH₂Cl₂ yielded 1 (7 mg) and C-3epi-wilsonine⁹ (5 mg); elution with CH_2Cl_2 -acetone (1:1) yielded 2 (10 mg), taiwanhomoflavone A² (9 mg), ginkagetin³ (8 mg), and apigenin³ (3 mg).

The heartwood (1.6 kg) was chipped and extracted with CHCl₃. The CHCl₃ extract (28 g) was chromatographed over silica gel, and elution with CH₂Cl₂-MeOH (1:1) yielded 3 (3 mg)

1,2-Didehydro-6,7-epoxy-3α,16,17-trimethoxyerythri**nan-8-one (1):** colorless powder; $[\alpha]_{D}^{28} + 11^{\circ}$ (*c* 0.104, CHCl₃); IR (KBr) ν_{max} 1689 cm⁻¹; UV (MeOH) λ_{max} (log ϵ) 280 (3.56), 225 (sh) (4.02), 206 (4.37) nm; ¹H NMR (CDCl₃, 400 MHz) δ 1.64 (1H, m, H_{α}-11), 1.99 (1H, m, H_{β}-11), 2.86 (1H, dd, J =15.6, 6.8 Hz, H_{α}-12), 3.18 (1H, dd, J = 15.6, 2.4 Hz, H_{β}-12), 3.21 (1H, dd, J = 12.4, 2.0 Hz, H_a-10), 3.31 (3H, s, OMe-3), 3.47 (1H, m, H_b-3). 3.83 (3H, s, OMe-16), 3.83 (1H, bs, H_b-7), 3.88 (3H, s, OMe-17), 4.46 (1H, td, J = 12.4, 3.2 Hz, H_{β}-10) 5.76 (1H, dd, J = 10.4, 2.4 Hz, H-1), 6.26 (1H, td, J = 10.4, 1.5 Hz, H-2), 6.64 (1H, s, H-18), 7.07 (1H, s, H-15); ¹³C NMR (CDCl₃, 100 MHz) & 27.7 (C-11), 36.3 (C-12), 38.7 (C-4), 41.2 (C-10), 55.9 (OMe-17), 56.0 (OMe-16), 56.2 (OMe-3), 59.7 (C-7), 65.9 (C-6), 67.6 (C-5), 73.6 (C-3), 112.3 (C-15), 115.3 (C-18), 124.8 (C-1), 128.3 (C-14), 133.3 (C-13), 136.8 (C-2), 146.4 (C-16), 148.0 (C-17), 167.9 (C-8); EIMS m/z 357 [M]+ (100), 342 (60), 341 (38), 326 (30), 298 (23), 286 (15); HREIMS m/z 357.1510 (calcd for C₂₀H₂₃NO₅, 357.1576).

Di(5,7,4"'-trihydroxy-2,3-dihydro-4',7"-dimethoxy-6"methyl-3',3""-biflavanyl)-6,6-tetraflavone (2): amorphous yellow solid; $[\alpha]_D^{28}$ +44° (*c* 0.114, MeOH); IR (KBr) ν_{max} 3158, 1642, 1607 cm⁻¹; UV (MeOH) λ_{max} (log ϵ) 208 (4.85), 278 (4.55), 329 (4.31) nm; (MeOH-NaOMe) 218, 286, 386 nm; (MeOH-AlCl₃) 210, 256, 285, 348, 386 nm; (MeOH-NaOAc) 217, 285, 331 nm; (MeOH-NaOAc-H₃BO₃) unchanged. ¹H NMR (C₅D₅N, 400 MHz), see Table 1; ¹³C NMR (C₅D₅N, 100 MHz), see Table 1; ESIMS m/z 1187 [M + Na + 2H]⁺ (100), 1164 [M + 2H]⁺ (6), 895 (18), 771 (18), 727 (24), 595 (23), 583 (51), 457 (15); ESIMS m/z 1187.0 [M + Na + 2H]⁺ (calcd for C₆₆H₅₂O₂₀Na $[M + Na + 2H]^+$ 1187.3).

5,6,8,9-Tetrahydro-1-hydro-4*H*-cyclopenta[*a*]-[1,3]dioxolo[4,5-*h*]pyrrolo[2,1-*b*][3]benzaepin-2(3*H*)-one (3): brownish powder; $[\alpha]_D^{28}$ +52° (*c* 0.03, MeOH); IR (KBr) ν_{max} 3422, 1718 cm $^{-1};$ UV (MeOH) λ_{max} (log $\epsilon)$ 212 (4.64), 288 (4.22) nm; ¹H NMR (C₅D₅N, 400 MHz) δ 1.82 (1H, m, H_α-6), 1.96 (1H, m, H_{α}-7), 2.00 (1H, m, H_{β}-7), 2.02 (1H, m, H_{β}-6), 2.83 (1H, d, J = 18.2 Hz, H_a-1), 3.80 (1H, d, J = 18.2 Hz, H_b-1), 3.20 $(1H, m, H_{\alpha}-8)$, 3.69 $(1H, m, H_{\beta}-8)$, 3.17 $(1H, m, H_{\alpha}-10)$, 3.67 (1H, m, H_{β} -10), 3.13 (1H, m, H_{α} -11), 3.20 (1H, m, H_{β} -11), 6.00 (1H, s, H-18), 6.07 (1H, s, H-18), 6.71 (1H, s, H-17), 7.32 (1H, s, H-14); ¹³C NMR (CDCl₃, 100 MHz) & 22.2 (C-7), 31.3 (C-11), 36.1 (C-6), 47.4 (C-1), 49.3 (C-8), 51.9 (C-10), 73.2 (C-5), 102.1 (C-18), 110.0 (C-17), 110.9 (C-14), 125.4 (C-13), 137.2 (C-4), 132.2 (C-12), 146.8 (C-16), 148.5 (C-15), 154.3 (C-3), 198.8 (C-2); EIMS m/z 299 [M]+ (79), 282 (23), 271 (12), 256 (100), 228 (72); HREIMS *m*/*z* 299.1155 (calcd for C₁₇H₁₇NO₄, 299.1157).

Tumor Cell Growth Inhibition Assays. A microassay for cytotoxicity was performed with the MTT (3-[4,5-dimethylthiazo-2-yl]-5-[3-carboxymethoxymethoxyphenyl]-2[4-sulfophenyl]-2H-tetrazolium bromide) assay.^{12,13} Briefly, $(1-3) \times 10^3$ cells/ 100 µL were seeded in 96-well microplates (Nunck, Roskilde, Denmark) and preincubated for 6 h to allow cell attachment. The cells were incubated with each drug for 6 days and then pulsed with 10 μ L of MTT (5 mg MTT/mL; Sigma, St. Louis, MO) and incubated for an additional 4 h at 37 °C. The microplates were read at 550 nm on a Multiskan photometer (MR5000; Dynatech, McLean, VA) after lysis of cells with 100 μL of 10% SDS (sodium dodecyl sulfate) in 0.01 M HCl. Control wells contained medium plus cells (total absorbance) or medium alone (background absorbance). Cell death was calculated as the percentage of MTT inhibition.

Human hepatomacellular carcinoma Hep 3B and Hep G2, human colorectal adenocarcinoma HT-29, and human breast adenocarcinoma MCF-7 cells were obtained from American Type Culture Collection (ATCC; Rockville, MD) and grown in DMEM,13,14 containing 10% FBS, 2 mM L-glutamine, 100 units/ mL penicillin, and 100 $\mu \mathrm{g/mL}$ streptomycin. For the microassay, the growth medium was supplemented with 10 mM HEPES (4-(2-hydroxyethyl)piperazine-1-ethanesulfonic acid) buffer pH 7.3 and incubated at 37 °C in a CO₂ incubator.

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Supporting Information Available: Portions of ¹H NMR signals of 2 in C₅D₅N (A) and CD₃OD (B) are shown in Figure S1. This material is available free of charge via the Internet at http://pubs.acs.org.

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